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T182 Chemistry: Matter and Change  
Name Date Class Name Date Class  
CHAPTER 9 STUDY GUIDE FOR CONTENT  
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Section 9.2 Naming Molecules In your  
textbook, read about how binary  
compounds and acids are named from  
their formulas.

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Skills Search for Context Clues

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## Answer Key

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### **Mastery Answers ...**

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CHAPTER 17 Name Date Class Study

Guide for Content Mastery Chemistry:

Matter and Change • Chapter 17 97

Reaction Rates Section 17.1 A Model for

Reaction Rates In your textbook, read

about expressing reaction rates and

explaining reactions and their rates. Use

each of the terms below just once to

complete the...

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STUDY GUIDE FOR CONTENT MASTERY

The Structure of the Atom section 4.1

Early Theories of Matter In your

textbook, read about the philosophers,

John Dalton, and defining the atom. For

each statement below, write true Or

false. 1. Ancient philosophers regularly

performed controlled experiments. 2.

### **atom-work-sheets - Mister Chemistry**

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17 Section 3.4 Elements and

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Compounds In your textbook, read about elements and compounds. Circle the letter of the choice that best completes the statement or answers the question.

1. A substance that cannot be separated into simpler substances by physical or chemical

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Name CHAPTER Date Class STUDY  
GUIDE FOR CONTENT MASTERY Section  
17.2 Factors Affecting Reaction Rates In  
your textbook, read about the factors  
that affect reaction rates (reactivity,  
concentra- tion, surface, area,  
temperature, and catalysts).

## **Livingston Public Schools / LPS Homepage**

CHAPTER 13 STUDY GUIDE FOR  
CONTENT MASTERY Name Date Class  
Study Guide for Content Mastery  
Chemistry: Matter and Change Chapter  
13 75 Section 13.2 Forces of Attraction  
In your textbook, read about forces of

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## Answer Key

attraction. Answer the following questions. 1. Ionic, metallic, and covalent bonds are examples of what type of forces? 2.

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### **Solved: CHAPTER 19 STUDY GUIDE FOR CONTENT MASTERY Earthqu ...**

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## Answer Key

Chemistry: Matter and Change • Chapter 3  
17 Section 3.4 Elements and Compounds  
In your textbook, read about elements and compounds. Circle the letter of the choice that best completes the statement or answers the question.

1. A substance that cannot be separated into simpler substances by physical or chemical

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